

## *Chemistry Sba Review Answers*





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Best Answer: Isotopes have the same proton/atomic number (the smaller number and on bottom) but different mass numbers (the larger number and on top). Basically chromium has a normal mass number of 52 and atomic number 24, so an isotope of it would be atomic number 24 (on bottom) and any number other than 52 on top (+or -), for example, 53, 51, etc. Hope that helped!

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### **Chemistry SBA Part 1**

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