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10—2 Review and Reinforcement (continued) 9. $0.0621 \text{ mol SS} = (Q \text{ } \$ \text{ gK}2\text{S } 10. 65.8\text{gMg} = \text{l. atoms Mg}$ Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NO₂ gas.

10-2 Review and Reinforcements

2 Answer.Key 2-1 Explore, page 5 ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed ... 17. distillation 18. A magnet could be used to separate them because iron is attracted to a magnet but

2 Answer - urbanbhs.weebly.com

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'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of a liquid placed in. a container, equilibrium is established when the Equid vaporizes at the same rate that vapor condenses. 7, NO, in reversible reactions at equilibrium,

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Stoichiometry Handout Answers.

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11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdrvption of matter coefficients subscripts mass

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11—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles Of magnesium are required to react with 2.0 mol of hydrochloric acid? The ... 17. Zinc reacts with H₂S₀4 to yield ZnS₀4 and hydrogen gas. How manv moles of hydrogen will

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